ECONOMIC ANALYSIS OF PRODUCING ALUMINA WITH LOW-GRADE BAUXITE(RED MUD) BY CALCIFICATION-CARBONIZATION METHOD

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Abstract

Calcification-carbonization is a new method of producing alumina with low-grade bauxite (red mud). Low-grade bauxite (red mud) are all converted into garnet hydration during calcification process, then garnet hydration are treated with CO₂ and the new structure red mud are gained by digestion. Main components of the new sturcture red mud are CaSiO₃ and CaCO₃, it can be used directly for cement industry and Bauxite resource recycling is carried out. A series of experiments were done by our team and confirmed that the new method was feasible, A/S of new structure red mud dropped to 0.21 and the content of sodium is under 0.2%. Process calculation software METSIM was used to design and simulate the new process of producing alumina with low-grade bauxite (red mud) by calcification-carbonization method, and its economic with series process and bayer process was analyzed.

Introduction

At present, Bayer process is widely adopted in the industry of producing alumina. By using the method of alumina production, there are two main problems. One is the low yield and low economic benefit in the process of producing alumina with alumina bauxite of low aluminum silicon ratio. Two is that a large amount of red mud which has high alkali content can not be directly used. Equilibrium phase of the leaching residue (red mud) in Bayer process is hydrated alumino silicate. In its production process, the loss of alumina and consumption of caustic alkali both increase with the higher silica content in minerals. The proportion of alumina and silica content loss relationship is 1:1, but the actual production is more than 1:1. Such as producing alumina with alumina bauxite of aluminum silicon ratio of 3, alumina theoretical maximum yield is only 66%, while the caustic consumption is up to more than 90 kg/t. Therefore, contradiction between low grade bauxite resources and production technology has become one of the main bottlenecks of China's alumina industry and Aluminum recycling economy development.

In Bayer process digestion of red mud dealkalization, from economic considerations, lime is a kind of effective removal of alkali additives. Aluminum industry in China has done a lot of research and practice of industrial production in the 70~80's in Lime-Bayer process [1-3]. But because equilibrium phase of digestion slag is hydrogarnet in the Lime-Bayer process, the A/S ratio is higher than that of acid sodium hydrated silicon aluminum, so the overall yield of the method is inferior to ordinary alumina Bayer process. At present, lime dosage is low in China's alumina production enterprises, at around 10 %. Sodium alkali content in dissolving slag is up to more than 4 %. The comprehensive utilization is difficult. In the Bayer de-alkali process, the lime is undoubtedly the most suitable additives. So as long as to find a simple and easy method of extracting alumina in the hydrate garnet, it can simultaneously reduce alumina alkali consumption of the red mud and improve overall yield, to achieve efficient resource recycling aluminum. After years of research, Northeastern University puts forward the calcification-carbonization method of processing low-grade bauxite (red mud) and other alumina materials [4-8]. In view of this new technology, this paper adopts METSIM flow calculation software to carry out the design and simulation, and compares it with series process and Bayer process on the economic benefits

1. Technology fundaments

This technology firstly converts bauxite or red mud into garnet hydration during calcification process, secondly treats garnet hydration with CO_2 to gain carbide slag which is mainly composed of calcium silicate, calcium carbonate and aluminum hydroxide during Carbonization process, and then obtains red mud of a new structure by low-temperature soluble aluminum. Main sturcture components of the new red mud are CaSiO₃ and CaCO₃, it can be used directly for cement industry and Bauxite resource recycling is carried out.

The calcification reaction and carbonization reaction are respectively shown in formula (1) and (2).

 $Na_2O{\cdot}Al_2O_3{\cdot}1.7SiO_2{\cdot}2.6H_2O{+}3CaO{+}2H_2O$

 $\rightarrow 3CaO \cdot Al_2O_3 \cdot 1.7SiO_2 \cdot 2.6H_2O + 2NaOH \\ 3CaO \cdot Al_2O_3 \cdot 1.7SiO_2 \cdot 2.6H_2O + 1.3CO_2 + H_2O \rightarrow$

0.64CaSiO₃+1.3CaCO₃+2Al(OH)₃

The dissolution reaction is shown in formula (3). Al(OH)₃+NaOH \rightarrow Na₂Al₂O₄+H₂O

Technological process of producing alumina with red mud by calcification-carbonization method is shown in figure 1.

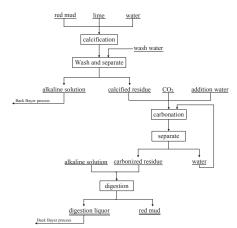


Figure 1 - Process flow diagram

2. METSIM Material Balance Calculation

2.1 Raw Database

According to the principle of calcification -carbonation reaction, composition of reactant and product were determined. In order to meet the needs of chemical reaction, the relevant elements and substances were selected in the METSIM database, which can provide physical, chemical and thermodynamic data. The operation interface was shown in figure 2.

A120	Dialuminum Oxide	GAS	XX 35	B672041	298	
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A1203	Dialuminum Trioxide	SLAG	XX 37	B672042	2327	
A1203	Dialuminum Trioxide	SOLID		B672044	298	
A1204H2	Al Oxide Monohydrate	SOLID	XX 67	BAK2027A	298	
A1206H6	Al Oxide Trihydrate	SOLID	XX 68	BAK2027B	298	
A12S105	Kyanite	SOLID	XX2420	B6773091	298	
A125105	Andalusite	SOLID	XX2421	B6773092	298	
A125105		SOLID	XX2422	B6773093	298	
A128105	Andalusite	SOLID	XX2449	BAK1 038	298	
A125105	Kyanite	SOLID	XX2450	BAK1039	298	
A128105	Sillimanite	SOLID	XX2451	BAK1 04 0	298	
A1251207	Aluminum Disilicate	SOLID	XX2448	BAK1037A	298	
A12S1209H4	Kaolinite	SOLID	XX2464	BAK2039A	298	
A12514012H2	Pyrophyllite element	SOLID	XX2446	1452423	298	
A14C3	Tetraluminum Tricarb	SOLID	XX 48	B6771312	298	
A16512013	Mullite elements	SOLID	XX2433	1452345	298	
A16 S12013	Mullite	SOLID	XX2452	BAK1041	298	
A16512013	Mullite	SOLID	XX2463	BAK2038	298	
A1 (OH) 3	Aluminum Hydroxide	SOLID	XX 57	BAK1 035A	298	Ŧ

Figure 2 - Selection interface of compound component

2.2 Establishment of Material balance calculation flow diagram

According to the process flow diagram of producing red mud by calcification-carbonization method, the METSIM calculation of material balance was shown in figure 3.

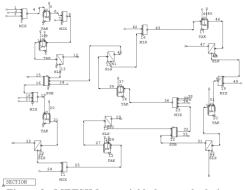


Figure 3 - METSIM material balance calculation process of producing red mud by calcification-carbonization method

The process includes the calcification process, three-stage carbonization process and the corresponding digestion process of carbide slag. In order to make the operation simple, the choices of equipments were mainly stirred tank, mixing tank solid-liquid separation equipment and so on.

The computation process of red mud in a place as raw material, and its main components were shown in table 1.

Table 1 - Composition of red mud

Composition	Al_2O_3	SiO ₂	Na	Fe	L.O.I	A/S
Content (%)	20.3	33.82	7.53	16.41	9.7	0.6

Calcification conditions: liquid- solid ratio 4:1, calcium-silicon ratio 2.5:1 (mass) (CaO:SiO₂=2.5:1), the reaction rate was 95%, reaction temperature 120 °C, system pressure 3Mp, reaction time 30min. Carbonization conditions: liquid-solid ratio 10:1, reaction temperature 120 °C, system pressure 1.2 Mp, the single reaction rate was 20%, reaction time 120min. Dissolution conditions: if using lye, 100 g/L NaOH, liquid- solid ratio 10:1, reaction rate was 100%, reaction time 90min. After settling separation

of each section, solids holdup underflow was 45%. **2.3 Input of Original Data**

Based on the above established calcification-carbonation reaction process and the test results, reaction equation was input and all the reaction efficiency were set in the unit operation of calcification, carbonization and dissolution reaction, which were shown in figure 4.

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Figure 4 - Input of reaction equation

3. Results and Discussion

In alumina plant with an annual output of 250000 tons dry red mud calculation, the results were as follows: water requirement in calcification process was 3511.24 Kg/min, CaO flow obtained by calcification reaction was 402.16 Kg/min, NaOH concentration in calcified slag was 16.12 g/l. The above results were in good agreement with the test results. According to the logistics calculation results, the economic benefit was analyzed, which was shown in table 3.

According to the factory annual emissions 250,000 tons of red mud, the economic benefits of this technology can reach more than 200 million yuan. If used for processing the existing red mud, economic benefits can be generated more than 40 billion yuan (by province stockpiles of 50 million tons of red mud calculation)

Conclusion

(1) Metsim software was successfully used in the process of producing alumina with low-grade bauxite (red mud) by calcification-carbonization method. The operation results had good agreement with the test results and it was reliable and accurate.

(2) Favorable economy benefit can be reached to treat low-grade bauxite (red mud) by using this method and it was better than the similar technology. Production cost was 2598 Yuan per ton alumina with alumina-silica ratio 3.29 and reduced by 400Yuan compared with traditional Bayer. Only core equipment for carbonization process was needed to add to Alumina production enterprises the method has vast extension prospect.

Table 2 - Reaction efficiency

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(c)

Main reaction	Calcification	Primary carbonization	Secondary carbonization	Third carbonization	Dissolution
Efficiency%	95	20	20	20	100

Table 3 - Product value and consumption obtained with processing one ton Bayer red mud by Calcification-Carbonization Method

Products and consumption of raw materials	Additional product yield and consumables usage, tons of red mud	Unit price, RMB / t	Single products price, RMB / t
Products – caustic alkali	0.05	2450	122.5
Products - alumina	0.18	2600	468
Products - calcium silica slag	1.4	200	280
Reduce red mud maintenance	1	50	50
costs			
Lime consumption	0.8	300	-240
Total			880.5

Acknowledgement

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